

Vacancy–Strain Coupling in Ordered TiO

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The structure of the partially ordered phase of TiO has been examined with a single crystal. While the space group is that first determined from powder X-ray diffraction and electron diffraction by Watanabe, Castle, Jostons & Malin [*Acta Cryst.* (1967). **23**, 307–313], ($B2/m$), the average structure is somewhat different. There are waves of Ti and O ion vacancies every third (100) plane in the [010] direction of the monoclinic structure. The amplitude of the wave is twice as large for the Ti ion vacancies as for the O ion vacancies. There is no appreciable concentration of interstitial Ti ions in either the ordered phase or the disordered (cubic) phase. The ordering temperature T_c is 1026 (2)°C. The monoclinic angle is directly proportional to the square of the order parameter associated with vacancy ordering. A mean-field theory is developed that includes vacancy–vacancy and vacancy–electron interactions, and vacancy–strain coupling. This theory explains the main features of the order–disorder transition in this material.

Introduction

The monoxides of the first transition series have structures which at first glance are all that of NaCl (*Strukturbericht* type *B1*). But there are now known to be many details which do not fit this simple description, such as interstitial metal ions, vacancy clusters and various ordered phases, often involving such clusters. For example, for TiO 14.5% of the anion and cation sites are vacant (Banus & Reed, 1970; Taylor & Doyle, 1970). Watanabe, Castle, Jostons & Malin (1967) and Watanabe, Terasaki, Jostons & Castle (1968, 1970) reported that these vacancies were ordered below $\sim 950^\circ\text{C}$ and the NaCl structure was distorted to a monoclinic system, space group $B2/m$ (No. 12). In particular, it was proposed that half of the Ti and half of the O ions are missing every third (110) plane in the cubic phase [equivalent to every third (100) plane in the monoclinic phase]. However, this conclusion was based primarily on only 26 superlattice powder reflections and we calculated the R factor for these to be 0.33 from this data. A more detailed study is presented here.

In a previous paper (Terauchi & Cohen, 1977), we pointed out that the diffuse intensity present above T_c implies important vacancy–vacancy and vacancy–electron interactions. But the distortion of the unit cell

when the vacancies order implies that vacancy–strain coupling is also important in this system below T_c .

In this paper, the temperature dependence of the intensity of superlattice reflections and the monoclinic angle have been measured at high temperatures. A theory is developed for the transition that includes the above mentioned interactions; this theory describes the essential features of the transition.

Procedures

Crystals were prepared in a purified argon atmosphere in a tri-arc furnace (Reed & Pollard, 1968). The starting materials were TiO_2 (99.9% pure) and Ti (99.9% pure). After melting, coarse-grained specimens were obtained, with grains of approximate size $1 \times 1 \times 9$ mm, which were quite suitable for X-ray studies. Such specimens were homogenized at 1200°C for 2 d. The crystal employed for determining the ordered structure was annealed at 950°C for 15 h after a 15 h cool from the higher temperature. This was the same temperature employed by Watanabe *et al.* (1967). This heat treatment was chosen because when the ordered phase forms and the structure changes from cubic to monoclinic there are different domains and, furthermore, there is twinning in each domain. If the specimen is fully ordered at lower temperatures the various domains are not equal in size.

All heat treatments were carried out in helium gas (99.9% pure) flowing at $7 \text{ cm}^3 \text{ min}^{-1}$, passed over Ti

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Table 1. *Parameters for TiO*

Space group: $B2/m$ (No. 12), $a_m = 9.340$ (5), $b_m = 5.860$ (4), $c_m = 4.141$ (1) Å, $\gamma_m = 107.553$ (6)° (weighted $\lambda = 1.5418$ Å).

Atom	Equipoint	Occupation factor	(000), $(\frac{1}{2}0\frac{1}{2}) +$			$B(\text{Å}^2)$	$\langle u^2 \rangle^{\frac{1}{2}}(\text{Å}) \times 10^2$
			x	y	z		
Ti(1)	2(<i>a</i>)	0.284 (20)	0	0	0	2.756	3.49
Ti(2)	2(<i>c</i>)	0.941 (10)	0	$\frac{1}{2}$	0	1.120	1.42
Ti(3)	4(<i>i</i>)	0.952 (10)	0.3360 (7)	0.1620 (13)	0	0.925	1.17
Ti(4)	4(<i>i</i>)	0.958 (8)	0.3396 (6)	0.6670 (14)	0	0.852	1.08
O(1)	2(<i>b</i>)	0.782 (11)	0	$\frac{1}{2}$	$\frac{1}{2}$	1.906	2.41
O(2)	2(<i>d</i>)	0.462 (35)	0	0	$\frac{1}{2}$	1.206	1.53
O(3)	4(<i>i</i>)	0.945 (12)	0.1765 (24)	0.3244 (47)	0	1.152	1.46
O(4)	4(<i>i</i>)	0.948 (10)	0.1642 (24)	0.8056 (44)	0	1.214	1.54

In disordered cubic phase: $B_0 = 1.105 \text{ Å}^2$, $B_{\text{Ti}} = 0.937 \text{ Å}^2$.

chips maintained at 650°C, and then through a trap at dry-ice temperatures. Cooling to room temperature involved turning off the furnace while maintaining the gas flow. At about 500°C, specimens were removed to a cold zone just outside the furnace. Crystals were checked for weight loss and any change in the position of a high-angle peak, to verify that the composition was unchanged.

For the X-ray measurements, unless otherwise noted, a modified GE XRD-5 diffractometer with a quarter circle was employed (Schwartz, Cohen & Morrison, 1964) with an incident-beam monochromator of pyrolytic graphite, singly bent to focus the beam vertically at the receiving slits. Cu $K\alpha$ radiation in point focus, and a proportional counter with a pulse-height analyzer set for ~90% acceptance of the Cu $K\alpha$ were employed. An 8 μm thick Ni filter in front of the detector reduced the background (largely Ti fluorescence) by one third, while reducing a Bragg peak by only 12%.

Both ω and θ - 2θ scans were used to sample reciprocal space, but all integrations were carried out in the θ - 2θ mode employing a computer control (Richesson, Morrison, Cohen & Paavola, 1971) and in the $\chi = 90^\circ$ position to avoid absorption corrections. The beam was smaller than the crystal at all positions, and the FWHM of a reflection in an ω scan was about $0.014 a_c^*$, where a_c^* is the parameter of the unit cell of reciprocal space of the disordered cubic phase. Receiving slits were opened until integrations were independent of their size. Data were corrected for surface roughness by measuring the Ti fluorescence *vs* 2θ and correcting for the difference in absorption of the incoming Cu $K\alpha$ and outgoing Ti fluorescence (Suortti, 1972).

For studies of order *vs* temperature a crystal was maintained in an atmosphere of helium purified as indicated above. The furnace (Gehlen, 1969) was covered with a Be hemisphere 380 μm thick. The temperature was controlled with a feedback system to

$\pm 2.0^\circ\text{C}$. (The thermocouples, made of platinum-platinum 13% rhodium, were checked at the boiling point of water.)

Results

The partially ordered phase

Indices in the monoclinic unit cell (m) are related to those in the cubic cell (c) by:

$$h_m = h_c + 2k_c, \quad k_m = h_c - k_c, \quad l_m = l_c. \quad (1)$$

The lattice parameters of the monoclinic phase were obtained from the split 060, 111, 002 and 420 reflections. These parameters are given in Table 1; the uncertainty is estimated from errors in estimating the peak positions and the uncertainty in the true zero of the diffractometer. The monoclinic angle γ (107.55°) is less than that between the corresponding directions $[1\bar{1}0]$ and $[120]$ in the cubic phase by 0.85° .^{*} Reflections were present for the diffraction symbol $112/mB$. The most symmetric space group $B2/m$ (No. 12) was chosen for least-squares refinement (Busing, Martin & Levy, 1962). The refined R parameter was

$$R = \sum_i (|F_i^{\text{obs}}| - |F_i^{\text{calc}}|) / F_i^{\text{obs}}. \quad (2)$$

The weighted R parameter was also calculated:

$$R_w = \left[\sum_i w_i (|F_i^{\text{obs}}| - |F_i^{\text{calc}}|)^2 / \sum_i w_i |F_i^{\text{obs}}|^2 \right]^{1/2}. \quad (3)$$

Here w_i was taken as the inverse of the net intensity above background. At first, the temperature factors were fixed at the overall values calculated in Watanabe *et al.* (1967), and the occupancies, conversion factor and positions were refined. Then temperature factors

^{*} As will be shown, γ is very sensitive to order. As Watanabe *et al.* (1967) in their study found $\gamma = 107.52^\circ$, their degree of order was essentially the same as that in our specimen (which is ~ 0.8 at 950°C).

Table 2. Observed and calculated superstructure structure factors

R index: 0.104; weighted R index: 0.116.

h	k	l	F_{obs}	F_{calc}	h	k	l	F_{obs}	F_{calc}
400			35.48	35.06	101			19.35	20.39
800			6.11	5.21	111			25.94	27.19
010			19.61	21.82	311			9.12	11.90
210			17.45	17.45	711			10.64	13.93
410			17.00	15.61	111			26.01	28.97
610			14.42	11.09	711			11.59	7.87
210			17.96	20.52	321			3.21	5.13
410			19.85	16.47	521			35.54	34.08
610			15.21	12.77	131			28.16	26.81
810			11.21	12.50	331			13.67	16.06
020			39.33	36.07	212			14.93	14.29
420			11.44	11.51	412			11.66	12.42
220			29.32	28.35	612			9.94	9.22
030			9.51	12.98	212			12.66	15.15
230			10.33	14.46	412			11.02	12.81
430			10.45	12.52	612			10.90	11.10
230			19.69	15.69	422			10.72	9.97
430			18.27	14.91	222			24.21	22.43
630			13.54	12.88	622			10.70	10.29
830			7.05	8.51	032			10.45	10.53
040			12.88	13.87	232			11.71	11.52
440			14.27	16.26	232			12.72	12.94
640			19.51	19.71	432			13.04	12.22
050			11.71	12.58	632			13.07	10.71
250			12.28	11.42	042			12.88	12.53

were refined with the other factors fixed. Isotropic temperature factors were assumed, one for each atom type in each equipoint.* Three cases were examined: all 63 superstructure peaks and 11 fundamental peaks, the 63 superstructure peaks alone, and superstructure peaks alone, eliminating 13 superstructure reflections that were difficult to separate from peaks because of other domains or twins. The results for the last case are given in Table 1 and the structure factors in Table 2. [For the first case the R index was 0.151 and $R_w = 0.124$, for the second $R = 0.164$, $R_w = 0.181$. Except for the second case, the R factors are what may be expected from the number of unknowns, number of

* Scattering factors, including the dispersion corrections, were taken from *International Tables for X-ray Crystallography* (1962).

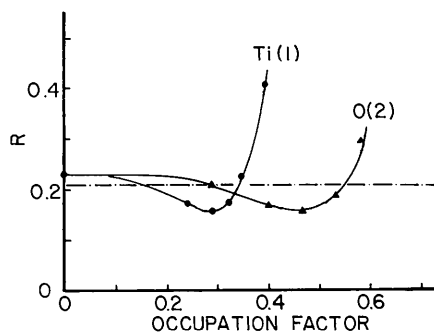


Fig. 1. R index vs occupation factor for Ti(1) and O(2) (see Table 1). The horizontal dashed line represents the 95% confidence limit following Hamilton (1965b).

peaks and the variation in intensity over the experiment (2.3%) following Hamilton (1965a).] In all three cases the parameters differed only by the sum of the standard deviations.

In Fig. 1, the occupancy factors of the Ti(1) and O(2) positions are plotted vs R . Also given is the 95% confidence limit (following Hamilton, 1965b) that the minimum values are different from the model of Watanabe *et al.* (1967). The structure is shown in Fig. 2. There are waves of vacancies in the [010] monoclinic direction, every third plane of O or Ti ions. But in contrast to Watanabe *et al.*'s (1967) result these waves do not have the same amplitude for the two ions, and there are some vacancies on planes parallel to the main wave.*

Summing the occupation factors (which were not held fixed in the refinement), the chemical formula is $\text{TiO}_{0.997}$ in agreement with the value determined by combustion to the highest oxide. The vacancy concentration is 16.2 at.% for O, 15.9 at.% for Ti, compared with the value of 14.5% determined for this composition (Banus & Reed, 1970) from the lattice parameter of the cubic phase, its density and combustion.

* The fully ordered phase probably contains vacancy waves with larger amplitudes, and fewer vacancies on planes parallel to the main wave. It was not possible to examine this state, because of the problems with domains mentioned earlier; this would be an interesting problem for studies by lattice imaging in the electron microscope.

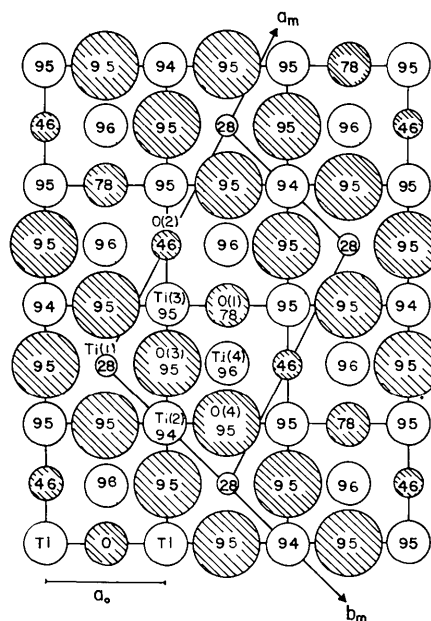


Fig. 2. The (001) plane in monoclinic TiO. Open circles and hatched circles show Ti and O ions respectively. a_m , b_m are axes of the monoclinic cell. Here, a_0 is the lattice constant of the cubic cell. The numbers in the circles indicate percent occupation, and the sizes of circles are proportional to occupation factors.

Refinement was also attempted for the other two space groups consistent with the extinctions, $B2$ and Bm . In both cases the y and z parameters, which are fixed at $\frac{1}{2}$ in equipoints, $2(b)$, (d) in $B2/m$, refined to the value of 0.500.

Tests for the presence of interstitial ions were also made; an interstitial occupancy was included and all parameters but the temperature factors were allowed to vary. The refinement converged to small negative values for the interstitial occupation. This result is surprising since such interstitials are known to occur in FeO (Smuts, 1966; Roth, 1960; Koch & Cohen, 1969), (Ni,Co)O (Stiglich, Whitmore & Cohen, 1973) and VO (Morinaga & Cohen, 1976). To confirm this further, one crystal was quenched from 1050°C (above the ordering temperature) by pulling it to the cold zone of the furnace from this temperature. Some 17 fundamental peaks were integrated and refinement including interstitials is indicated in Fig. 3. (The temperature factors for Ti and O in this disordered phase are given in Table 1.) Again, there is not clear indication of interstitials.

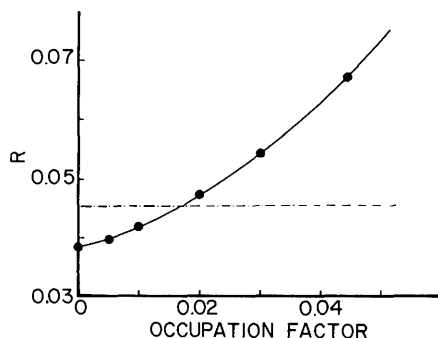


Fig. 3. R index vs Ti interstitial occupation factor in TiO. The horizontal dashed line represents 95% confidence limits, following Hamilton (1965b).

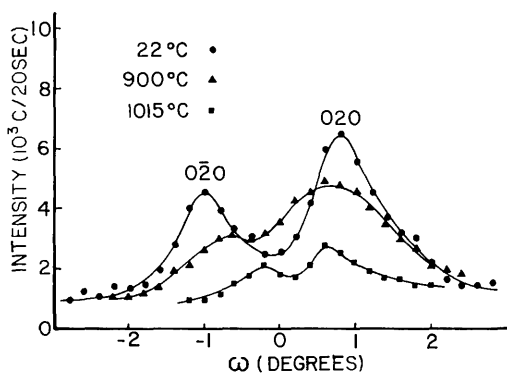


Fig. 4. ω scans near the 020 reflection in TiO at various temperatures.

Order vs temperature

The intensity distribution in an ω scan near the 020 monoclinic reflection is shown in Fig. 4 for three temperatures. The reflections 020 and 020 from two domains can be seen. The intensity decreases with increasing temperature and the splitting decreases. (It was found that equilibrium at each temperature required ~ 2 h.) The peak intensity of the 020 reflection vs temperature is given in Fig. 5. The 020–020 splitting, $\delta\omega = 2\delta\gamma$ ($\delta\gamma \equiv 90^\circ - \gamma_c$, γ_c being the pseudo-cubic angle) is shown vs temperature in Fig. 6. From these two figures it is clear that the splitting and superlattice intensity are proportional and hence splitting is proportional to the square of the order parameter.

The transition temperature T_c is 1026 (2)°C, 75°C above the temperature for continuous ordering (Terauchi & Cohen, 1977).

Discussion

TiO is the first monoxide of the 3d transition elements that has been shown *not* to contain ions in interstitial

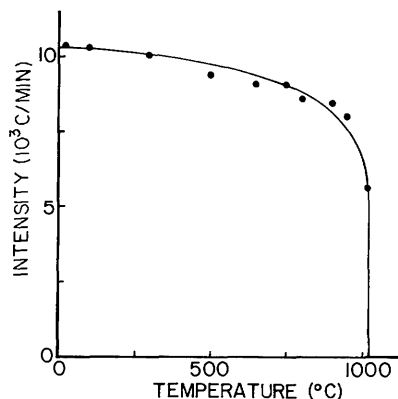


Fig. 5. Intensity of the 020 monoclinic reflection vs temperature.

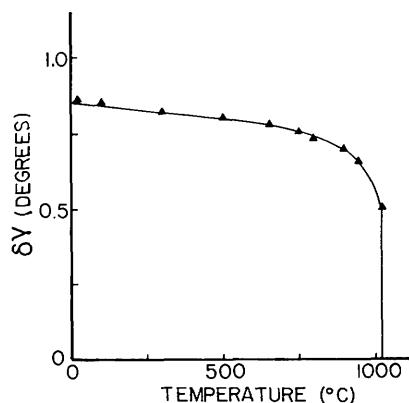


Fig. 6. $\delta\gamma = (90^\circ - \gamma)$ in TiO vs temperature.

positions of the rock-salt structure. When the vacancies order, the structure is monoclinic and the average structure contains waves of vacancies every third (100) plane of the monoclinic phase, but there are vacancies on the planes in between as well. These 'waves' are somewhat different for the Ti and O ions, but in what follows we shall ignore this difference in a first attempt to understand the various interactions in this system that lead to the monoclinic distortion below T_c . We thus assume that there is one vacancy position (not a separate one on each sublattice) with average concentration c .

In a previous paper (Terauchi & Cohen, 1977) we have shown that the Hamiltonian for this system can be written in a mean-field approximation in terms of pair interactions and vacancy concentrations at a site c_i .

$$H = \frac{1}{2} \sum_{ij} J_{ij} (c_i - c)(c_j - c) + \frac{1}{\sqrt{N}} \sum_{j\mathbf{q}} G_{\mathbf{q}} (c_j - c) \rho_{\mathbf{q}} \exp(i\mathbf{q} \cdot \mathbf{r}_{ij}). \quad (4)$$

Here the first term represents vacancy-vacancy interactions with energy J_{ij} for sites i and j separated by \mathbf{r}_{ij} . The second term represents a vacancy-electron interaction, the exponent being a charge-density wave of wave vector \mathbf{q} and amplitude $\rho_{\mathbf{q}}$. Writing the Fourier transformations of J_{ij} as $J_{\mathbf{q}}$ and with the transform of $\sigma_i \equiv c_i - c$ as $\sigma_{\mathbf{q}}$:

$$H = \frac{1}{2} \sum_{\mathbf{q}} J_{\mathbf{q}} \sigma_{-\mathbf{q}}^* \sigma_{\mathbf{q}} + \sum_{\mathbf{q}} G_{\mathbf{q}} \sigma_{\mathbf{q}} \rho_{\mathbf{q}} = \frac{1}{2} \sum_{\mathbf{q}} J_{\text{eff}}(\mathbf{q}) \sigma_{-\mathbf{q}}^* \sigma_{\mathbf{q}}, \quad (5)$$

where we have assumed that $\rho_{\mathbf{q}}$ is proportional to $\sigma_{\mathbf{q}}$ as a result of screening. (TiO is a semiconductor.) The effective potential $J_{\text{eff}}(\mathbf{q})$ represents the difference in the (transformed) potentials, vacancy-vacancy minus vacancy-electron.

In the ordered phase, $\langle \sigma_{\mathbf{q}_0} \rangle \equiv \sigma$. That is, the spatial average of $\sigma_{\mathbf{q}}$ at the value of $\mathbf{q} = \mathbf{q}_0$, a superlattice position, does not vanish; it is an order parameter. The free energy per atom with $J_{\text{eff}} \equiv J_{\text{eff}}(\mathbf{q}_0)$ can then be written

$$F/N = \frac{1}{2} J_{\text{eff}} \sigma^2 + k_B T \left[\frac{1}{2} (1 + \sigma) \log \frac{1}{2} (1 + \sigma) + \frac{1}{2} (1 - \sigma) \log \frac{1}{2} (1 - \sigma) \right]. \quad (6)$$

Here the second term is the entropy, which is assumed to be that of an Ising system. However, this expression does not include the distortion that has been shown to be coupled to ordering in this system. To include this coupling, we now expand the effective potential or ordering energy in terms of the strains, e :

$$J_{\text{eff}} = J_{\text{eff}}^0 + \sum_{\alpha, \beta} \frac{\partial J_{\text{eff}}}{\partial e_{\alpha\beta}} e_{\alpha\beta}, \quad (7)$$

where α, β represent coordinate axes, x, y, z . Also the

temperature of continuous order is defined (from Terauchi & Cohen, 1977) as:

$$J_{\text{eff}}^0 + k_B T_0 = 0. \quad (8)$$

The distortion $\Delta\gamma = 90^\circ - \gamma_c$ between the a and b axes is equivalent to a shear strain, e_{xy} . If other strain components are ignored for simplicity, the free energy may be written as:

$$F/N = \frac{1}{2} J_{\text{eff}}^0 \sigma^2 + \frac{1}{2} J'_{\text{eff}} e_{xy} \sigma^2 + \frac{v}{2} C_{66} e_{xy}^2 + K_B T \left[\frac{1}{2} (1 + \sigma) \log \frac{1}{2} (1 + \sigma) + \frac{1}{2} (1 - \sigma) \log \frac{1}{2} (1 - \sigma) \right]. \quad (9)$$

Here v is the volume of the unit cell and C_{66} is the elastic constant for the shear e_{xy} . The third term represents the shear strain energy. The temperature dependence of both the order parameter and σ and e_{xy} are then determined from the stability conditions;

$$\frac{\partial F}{\partial \sigma} = \frac{\partial F}{\partial e_{xy}} = 0, \quad (10)$$

which results in the following pair of equations:

$$e_{xy} = \frac{-J'_{\text{eff}}}{2C_{66}v} \sigma^2, \quad (11a)$$

$$J_{\text{eff}}^0 \sigma - \frac{J_{\text{eff}}^2}{2C_{66}v} \sigma^3 + \frac{1}{2} k_B T \ln \left(\frac{1 + \sigma}{1 - \sigma} \right) = 0. \quad (11b)$$

Equation (11a) shows that the strain is proportional to the square of the order parameter and hence to the intensity of a superlattice reflection, as is indicated by the experimental data. The temperature dependence of σ (and hence the strain) is determined by (11b). The latter equation is most readily solved by defining two quantities:

$$X_1 = -J_{\text{eff}}^0 \sigma + \frac{J_{\text{eff}}^2}{2C_{66}v} \sigma^3, \quad (12a)$$

$$X_2 = \frac{1}{2} k_B T \ln \left(\frac{1 + \sigma}{1 - \sigma} \right). \quad (12b)$$

The value of J_{eff}^0 is 70.3 meV (Terauchi & Cohen, 1977). Therefore, σ can be plotted *vs* X_1 from (12a) for various values of $J_{\text{eff}}^2/2C_{66}v \equiv a$. Where this curve intersects σ *vs* X_2 for a given t/T_0 determines the value(s) of σ . The results are shown schematically in Fig. 7. When $a = 0$, *i.e.* $J'_{\text{eff}} = 0$ (no strain coupling), σ *vs* T is continuous as is typical for a second-order phase transition. The temperature where $\sigma = 0$ is T_0 . However, for $a = 1.0 k_B T_0$ there are two solutions for σ indicating a first-order phase transition at $T_c = 1.21 T_0$. Thus the first-order nature of the transition in TiO appears to be explained by vacancy-strain coupling.

In Fig. 8, the data on order (peak intensity) and strain are shown *vs* temperature. The best fit to the data is obtained with $a = 0.581 k_B T_0$ which was chosen to give the value of $T_c = 1026^\circ\text{C}$ found in this study and the known $T_0 = 951^\circ\text{C}$ (Terauchi & Cohen, 1977), *i.e.* T_c/T_0 is known experimentally to be 1.062. The intensity ratio at T_c is predicted to be 0.46, compared with the observed value of 0.57, and the actual strain at T_c , $\Delta\gamma = 0.51^\circ$, compared with the predicted value $\Delta\gamma = 0.39^\circ$.

Inserting (11a) into (9):

$$F = \frac{1}{2}J_{\text{eff}}^0\sigma^2 - \frac{1}{4}a\sigma^4 + \text{entropy term.} \quad (13)$$

The coefficient of the fourth power of σ is negative, since $a > 0$. Thus, the free energy has a secondary minimum at a positive value of σ in the range $T_0 < T < T_c$, as expected for a first-order transition. It is this fourth-power term in σ that is causing the first-order

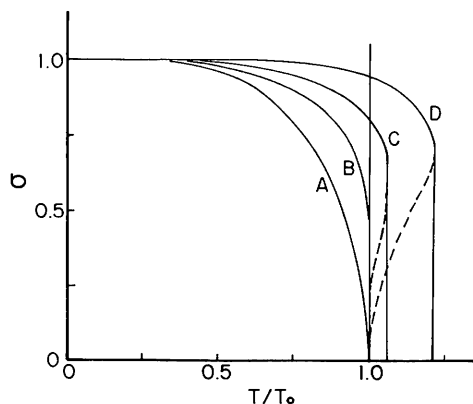


Fig. 7. Order parameter, σ , *vs* T/T_0 for various values of the constant, a : (A) $a = 0$, (B) $a = 0.333 k_B T_0$, (C) $a = 0.581 k_B T_0$, (D) $a = 1.0 k_B T_0$.

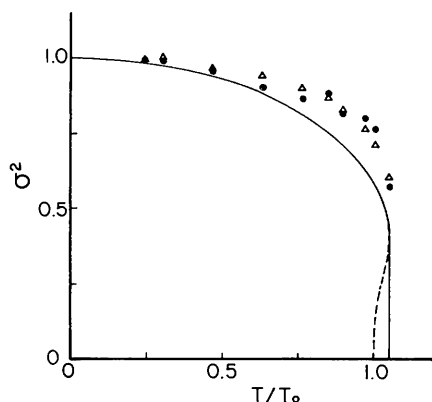


Fig. 8. Actual strain (triangles) and superlattice intensity (closed circles) *vs* T and the theoretical curve σ^2 for $a = 0.581 k_B T_0$.

transition (rather than the third-order term which is usually thought to be important in such cases). Also the reason for the appearance of domains is clear; the same minimum occurs at $\pm\sigma$.

No evidence was found in this study made at high temperatures for the transition phase, reported by Watanabe *et al.* (1968, 1970); they employed quenched specimens.

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